Lesson Plan of 1 st Semester			
Name of the Teacher: Hemant Kumar		Class: B.Sc. (N.M.)	Subject: Chemistry
16-08-2022 To 20-08-2022	Review of Bohr's theory and its limitations, dual behaviour of matter and radiation, de Broglie's relation, Heisenberg uncertainty principle		
22-08-2022 To 27-08-2022	Hydrogen atom spectra, what is Quantum Mechanics? Time independent Schrodinger equation and meaning of various terms in it. Significance of ψ and ψ^2 , Schrodinger equation for hydrogen atom		
29-08-2022 To 03-09-2022	Radial and angular parts of the hydrogenic wave functions and their variations 1s, 2s, 2p, 3s, 3p and 3d orbitals (only graphical representation) Radial and angular nodes and their significance, Radial distribution functions and concept of the most probable distance with special reference to 1s and 2s atomic orbitals		
05-09-2022 To 10-09-2022	Significance of quantum numbers, orbital angular momentum and Quantum numbers ml and ms, shapes of s, p and d atomic orbitals, nodal planes. Discovery of spin, spin quantum number (s) and magnetic spin quantum number (ms)		
12-09-2022 To 17-09-2022	Review of ionic bonding: General characteristic and energy consideration in ionic bonding, Lattice energy and solvation energy and their importance in the context of stability and solubility of ionic compounds.		
19-09-2022 To 24-09-2022	Statement of Born-Lande equation for calculation of lattice energy, Born-Haber cycle and its applications, polarizing power and polarizability, Fajan's rules, ionic character in covalent compounds bond movement dipole moment and percentage ionic character		
26-09-2022 To 01-10-2022	Covalent bonding: V molecules and ions or suitable example of	B approach - shape the basis of VSEPR	es of some inorganic and hybridization with laner, square planar,
3-10-2022 To 8-10-2022	bonding MOs and combinations of atom orbitals	their characteristic f mic orbitals, non-bo	nod, bonding and anti for s-s, s-p and p-p nding combination of
10-10-2022 То 15-10-2022	MO treatment of homonuclear diatomic molecules of 1 st and 2 nd periods and hetronuclear diatomic molecules such as CO, NO and NO ⁺ , comparison of VB and MO approaches		
17-10-2022 To 21-10-2022	Postulates of Kinetic	theory of gases and de ion of real gases f	erivation of the Kinetic from ideal behaviour,

22-10-2022 To 30-10-2022	Van der Waals equation of state for real gases, Boyle temperature, critical phenomena, critical constants and their calculation from Van der Waals equation, Andrews isotherm of CO ₂
31-10-2022 To 05-11-2022	Maxwell Boltzmann distribution law of molecular velocities and molecular energies and their importance, temperature dependence of these distributions. Most probable, average and root mean square velocities.
07-11-2022 To 12-11-2022	Collision cross section, collision number, collision frequency, collision diameter and mean free path of molecules. Viscosity of gases and effect of temperature and pressure on coefficient of viscosity.
14-11-2022 To 19-11-2022	Liquids: Surface tension and its determination using Stalagmometer, Viscosity of a liquid and determination of coefficient of viscosity using Ostwald's viscometer, effect of temperature on surface tension and coefficient of viscosity of liquid
21-11-2022 To 26-11-2022	Forms of solids, symmetry elements, unit cells, crystal systems, Bravais lattice types and identification of lattice planes.
28-11-2022 To 03-12-2022	Law of crystallography - Law of constancy of interfacial angle, law of rational indices.
05-12-2022 To 10-12-2022	X-ray diffraction by crystals, Bragg's law, structure of NaCl, KCl and CsCl, defects in crystals. Glasses and liquid crystals.
12-12-2022 То 17-12-2022	Revision